117.4 (2)

78.7 (3)

71.5 (3)

-62.6 (4)

$w = 1/[\sigma^2(F_o^2) + (0.0228P)^2]$	Scattering factors from
+ 0.4177 <i>P</i>]	International Tables for
where $P = (F_o^2 + 2F_c^2)/3$	Crystallography (Vol. C)

NI—C12	1.389(3)	N2—C16	1.316 (3)
NI—C16	1.410(3)	N2-C15	1.379 (3)
N1—C14	1.410(3)	C14—C15	1.407 (4)
C12—N1—C16	121.8 (2)	N2C15C8	127.5 (2)
CI2-NI-CI4	133.5 (2)	N2-C15-C14	112.1 (2)
C16—N1—C14	104.7 (2)	C8-C15-C14	120.4 (2)
C16—N2—C15	104.3 (2)	N2-C16-N1	114.1 (2)
CII—CI4—CI5	121.0(2)	N2-C16-C7	128.5 (2)

134.1 (3)

104.9(2)

-82.9(3)

60.7 (4)

-60.0(4)

Table 1. Selected geometric parameters (Å, °)

H atoms were placed geometrically and thereafter allowed to ride on their parent atoms.

NI-C16-C7

C3-C4-C5-C13

C2-C1-C12-C13

C4-C5-C13-C12

Data collection: *DIF*4 (Stoe & Cie, 1992*a*). Cell refinement: *DIF*4. Data reduction: *REDU*4 (Stoe & Cie, 1992*b*). Program(s) used to solve structure: *SHELXS*86 (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL*93 (Sheldrick, 1993). Molecular graphics: *XP* (Siemens, 1994). Software used to prepare material for publication: *SHELXL*93.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BM1233). Services for accessing these data are described at the back of the journal.

References

C11-C14-N1

C15-C14-N1

C12-C1-C2-C3

C1-C2-C3-C4

C2-C3-C4-C5

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1,2,3,4-Tetrahydrobenzimidazo[2,1-*b*]quinazoline

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Abstract

The title compound, $C_{14}H_{13}N_3$, was identified as one of four possible isomers from a condensation reaction. Three of the four rings are coplanar within 0.026 Å. The molecules stack in pairs across inversion centres, with a shortest $C \cdots C$ distance of 3.373 (3) Å.

Comment

Chemically synthesized purine analogues find numerous applications in clinical medicine and medical research. The pharmacological approach involves analogues in which the heterocyclic ring system has been modified so as to induce toxic effects when the analogue is incorporated into specific cell constituents. As part of our program directed towards the synthesis of purines and other antimetabolites (Elgemeie *et al.*, 1996, 1997, 1998), we have recently reported various successful approaches to syntheses of purine, pyrimidine and folic acid analogues. Derivatives of these ring systems are of interest as antimetabolites in biochemical reactions (Elgemeie *&* Fathy, 1995; Elgemeie & Hussain, 1994; Elgemeie *et al.*, 1994).

We report here a novel one-pot synthesis of a purine analogue by reaction of the sodium salt of 2-(hydroxymethylene)-1-cyclohexanone, (3), with 2-aminobenzimidazole, (1) (see scheme below). The reaction between (1) and (3) in the presence of piperidine acetate gives an adduct for which four isomeric structures, (4)–(7), are possible. Initial nucleophilic attack by the amino group at the carbonyl carbon could be followed by cyclization and elimination of water to give compound (5) or (7); alternatively, attack of the amino group at the formyl group followed by cyclization leads to compound (4) or (6). Spectroscopic methods did not identify the product unambiguously and therefore an X-ray structure determination was carried out.



The structure was confirmed as (7) by X-ray methods (Fig. 1), with displacement parameters, bond lengths and located H atoms providing unambiguous evidence. The tricyclic ring system incorporating atoms N1-N3 and C5–C14 is planar (mean deviation 0.026 Å); atoms C1-C4 of the saturated ring deviate from this plane by 0.032(3), 0.697(4), 0.062(3) and 0.166(3)Å, respectively, all on the same side.





Bond lengths and angles are as expected; the major angle deviations from ideal values are associated with the five-membered ring [e.g. C14—N3—C12 104.6(2) and C8-C13-N2 132.1 (2)°]. A search of the Cambridge Structural Database (Allen & Kennard, 1993) revealed no other examples of the same tetraannular system: the corresponding compound with an aromatic rather than a saturated ring, i.e. 3,4-dihydro-2Hbenzimidazolo[2,3-b]quinazoline, was investigated by Molina et al. (1994) but was disordered.

The molecules pack in stacked pairs related by inversion centres; the shortest $C \cdots C$ contact is one of 3.373 (3) Å involving $C12 \cdot \cdot \cdot C13(1-x, 1-y, 1-z)$.

$C_{14}H_{13}N_3$

Experimental

A solution of 2-aminobenzimidazole [(1); 1.33 g, 0.01 mol], the sodium salt of 2-(hydroxymethylene)-1-cyclohexanone [(3); 1.47 g, 0.01 mol], and piperidine acetate (1 ml) in water (50 ml) and ethanol (50 ml) was refluxed for 15 min. Acetic acid (1.5 ml) was added to the hot solution. The precipitated solid was collected by filtration and crystallized from benzene in 65% yield (m.p. 504 K).

Mo $K\alpha$ radiation

Cell parameters from 52

 $\lambda = 0.71073 \text{ Å}$

reflections

 $\theta = 10.0 - 11.5^{\circ}$

T = 143(2) K

Needle

Yellow

 $\mu = 0.085 \text{ mm}^{-1}$

 $0.6 \times 0.2 \times 0.2$ mm

Crystal data

 $C_{14}H_{13}N_3$ $M_r = 223.27$ Monoclinic $P2_1/n$ a = 6.4542 (14) Åb = 12.388(3) Å c = 13.501 (3) Å $\beta = 93.20(2)^{\circ}$ V = 1077.8 (4) Å³ Z = 4 $D_x = 1.376 \text{ Mg m}^{-3}$ D_m not measured

Data collection

Stoe Stadi-4 diffractometer	$\theta_{\rm max} = 25.04^{\circ}$
ω/θ scans	$h = 0 \rightarrow 7$
Absorption correction: none	$k = -14 \rightarrow 9$
3666 measured reflections	$l = -16 \rightarrow 16$
1913 independent reflections	3 standard reflections
1320 reflections with	frequency: 60 min
$I > 2\sigma(I)$	intensity decay: none
$R_{\rm int} = 0.032$	

Refinement

N

Refinement on F^2 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.36 \ {\rm e} \ {\rm \AA}^{-3}$ $R[F^2 > 2\sigma(F^2)] = 0.051$ $\Delta \rho_{\rm min} = -0.21 \ {\rm e} \ {\rm \AA}^{-3}$ $wR(F^2) = 0.126$ S = 1.047Extinction correction: none 1913 reflections Scattering factors from International Tables for 154 parameters H atoms: see below Crystallography (Vol. C) $w = 1/[\sigma^2(F_o^2) + (0.0462P)^2]$ + 0.4705P] where $P = (F_o^2 + 2F_c^2)/3$

Table 1. Selected geometric parameters (Å, °)

	-		
N1—C11	1.322 (3)	N3-C12	1.318 (3)
N1-C12	1.352 (3)	N3-C14	1.390(3)
N2—C9	1.366 (3)	C9—C10	1.357 (3)
N2-C13	1.393 (3)	C10-C11	1.432 (3)
N2—C12	1.397 (3)	C13—C14	1.415 (3)
C11—N1—C12	117.8 (2)	N1-C12-N2	120.7 (2)
C9-N2-C13	131.7 (2)	C8-C13-N2	132.1 (2)
C9-N2-C12	121.5 (2)	C8-C13-C14	123.8 (2)
C13-N2-C12	106.79 (18)	N2-C13-C14	104.1 (2)
C12-N3-C14	104.55 (19)	N3-C14-C5	130.3 (2)
N3-C12-N1	126.3 (2)	N3-C14-C13	111.5 (2)
N3-C12-N2	113.0(2)	C5-C14-C13	118.2 (2)
C10-C1-C2-C3	54.1 (3)	C2-C3-C4-C11	36.7 (3)
C1 - C2 - C3 - C4	-59.8(3)		

H atoms were placed geometrically and allowed to ride on their parent atoms. Residual electron density of $ca 0.3 \text{ e} \text{ Å}^{-3}$ near the C2 atom may indicate a very small degree of disorder.

Data collection: *DIF4* (Stoe & Cie, 1992*a*). Cell refinement: *DIF4*. Data reduction: *REDU4* (Stoe & Cie, 1992*b*). Program(s) used to solve structure: *SHELXS*86 (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL*93 (Sheldrick, 1993). Molecular graphics: *XP* (Siemens, 1994). Software used to prepare material for publication: *SHELXL*93.

We thank the Deutsche Forschungsgemeinschaft for supporting our collaboration, the Fonds der Chemischen Industrie for financial support and Mr A. Weinkauf for technical assistance.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BM1232). Services for accessing these data are described at the back of the journal.

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A Triclinic Polymorph of Hexaphenylcyclotrisiloxane

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Abstract

In triclinic crystals of 2,2,4,4,6,6-hexaphenylcyclotrisiloxane, $C_{36}H_{30}O_3Si_3$, the endocyclic C—C bond angles in the phenyl groups at the C atoms bound to silicon are reduced to about 117°. The other geometric parameters do not show any significant differences from expected values.

Comment

Crystals of the title compound, (I), a large scale precursor in silicone chemistry, were first prepared and investigated by Hyde et al. (1947). Two kinds of crystals were optically separated and were characterized as orthorhombic and triclinic polymorphs by X-ray powder diffraction. The structure of the orthorhombic polymorph was later determined by Bokii et al. (1972) and redetermined with higher precision by Tomlins et al. (1985). The redetermination revealed small but significant distortions of the phenyl rings from their ideal geometry. With the intention of generating diphenylsilanediol by reaction of dichlorodiphenylsilane with water in toluene in the presence of *tert*-amyl alcohol (Burkhard, 1945), hexaphenylcyclotrisiloxane, (I), was formed after the solution had been stored for more than two weeks at room temperature. The solid was recrystallized from warm 2-butanone and chloroform as large colorless blocks.



The present structure analysis of the triclinic polymorph also shows some significant distorsions of the phenyl rings, most significantly, the decrease of the C—C—C bond angles to $117.0 (4)^{\circ}$ at those phenyl C atoms bound at silicon. A similar distortion observed in the structure of 2,2-diphenyl-2-sila-1,3,4-trihydro-